

Artigo

Cytotoxic Activity of Chemical Constituents and Essential Oil from the Leaves of *Leonotis nepetifolia* (Lamiaceae)

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Atividade Citotóxica de Constituintes Químicos e Óleo Essencial das Folhas de *Leonotis nepetifolia* (Lamiaceae)

Resumo: O câncer é um problema de saúde que afeta grande parte da população mundial. Nesse contexto, vários grupos de pesquisa têm investigado moléculas com maior eficiência e menores efeitos colaterais. A espécie *Leonotis nepetifolia* é um arbusto pertencente à família Lamiaceae com atividade citotóxica relatada na literatura. Este trabalho descreve o isolamento dos compostos hentriacontano, palmitato de fitila, estigmasterol glicosídeo, 6,7-dimetoxi-5,3',4'-trihidroxi-flavona, apigenina-7-O-glicosídeo e luteolina-7-O-glicosídeo a partir de extratos e a composição química do óleo essencial das folhas de espécimes brasileiras de *L. nepetifolia*, além das suas atividades citotóxicas *in vitro*. Os compostos foram identificados por uma série de métodos espectroscópicos e espectrométricos, principalmente RMN (1D e 2D) e CG-EM, bem como por comparação com dados da literatura. A atividade citotóxica de compostos isolados e óleo essencial foi realizada utilizando o ensaio de brometo de 3-(4,5-dimetiltiazol-2-il)-2,5-difeniltetrazólio (MTT), contra linhagens de células tumorais HCT-116 (côlon humano) e SF-295 (glioblastoma). O álcool 1-octen-3-ol foi o composto majoritário do óleo essencial e os compostos hentriacontano, palmitato de fitila, estigmasterol glicosídeo, apigenina-7-O-glicosídeo e luteolina-7-O-glicosídeo foram descritos pela primeira vez nesta espécie. Todos os compostos testados e o óleo essencial apresentaram baixa atividade citotóxica para as linhagens celulares testadas, sugerindo que outros estudos fitoquímicos devam ser conduzidos para a descoberta de outros compostos responsáveis pela atividade citotóxica da espécie.

Palavras-chave: Produtos naturais; química de produtos naturais; plantas medicinais; Caatinga.

Abstract

Cancer is a health problem affecting a large part of the world population. In this context, several research groups have investigated molecules with higher efficiency and lower side effects. The species *Leonotis nepetifolia* is a shrub belonging to the Lamiaceae family with cytotoxicity activity reported in literature. This paper describes the isolation of compounds hentriacontane, phytol palmitate, stigmasteryl glucoside, 6,7-dimethoxy-5,3',4'-trihydroxyflavone, apigenin-7-O-glucoside and luteolin-7-O-glucoside from extracts and the chemical composition of essential oil from the leaves of Brazilian *L. nepetifolia* species in addition to its *in vitro* cytotoxic activities. All compounds were identified by a series of spectrometric and spectroscopic methods, mainly NMR (1D and 2D) and GC-MS, as well as by comparison with literature data. The cytotoxic activity of isolated compounds and essential oil was performed using the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay, against tumor cell lines HCT-116 (human colon) and SF-295 (glioblastoma). The alcohol 1-octen-3-ol was the majoritary compound of the essential oil and the compounds hentriacontane, phytol palmitate, stigmasteryl glucoside, apigenin-7-O-glucoside and luteolin-7-O-glucoside were described for the first time in this species. All compounds tested and essential oil showed low cytotoxic activity for the cell lines tested, suggesting that other phytochemical studies should be conducted for the discovery of compounds responsible by cytotoxic activity of the species.

Keywords: Natural products; chemistry of natural products; medicinal plants; Caatinga.

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Cytotoxic Activity of Chemical Constituents and Essential Oil from the Leaves of *Leonotis nepetifolia* (Lamiaceae)

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1. Introduction

Cancer is a disease characterized by uncontrolled cell growth and division.

Nowadays, cancer is one the major death causes in the world. Only in 2012, 15 % of deaths worldwide were attributable to this disease and the reports have shown which approximately 17 million cancer deaths per

year might occur by 2030.¹⁻⁴ Chemotherapy is a major treatment for cancer, however, it is very expensive and shows pronounced side effects. These problems show the necessity in discovery of new drugs able to overcome the limitations of the current therapies.¹⁻⁵

The use of natural products for treatment of diseases is as long as the history of humanity. The World Health Organization (WHO) reported that about 80 % of the world's population still relies on traditional system of medicine. Until today, natural products are a rich source for discovery of new drugs for the treatments of many diseases including cancer, because the diversity of compounds produced by plants, microorganisms, marine organisms and others.⁶⁻⁷

Leonotis nepetifolia is an African herbaceous plant belonging to Lamiaceae family and widely distributed in southern India and America. In Brazil, this species is popularly known as “cordão-de-São Francisco”, “cordão-de-frade” and “rubim” and in the folk medicine, is used to treat stomach ulcers and as antidiarrheal, anti-inflammatory, expectorant, anti-asthmatic, antipyretic, digestive and sedative.⁸⁻⁹

Reports involving this species have showed the presences of fixed oils, essential oils, terpenoids, saponins, flavonoids, steroids, alkaloids, iridoids and coumarins.^{8; 10-19} A previous study developed by our research group showed the cytotoxic activity of *L. nepetifolia* in human and murine cell lines.²⁰ Thus, the aim of this study was to investigate the cytotoxic potential of the isolated compounds and the essential oil from a *L. nepetifolia* Brazilian specimen.

2. Experimental

2.1. General experimental procedures

The chemical analysis of *L. nepetifolia* (essential oil and compounds 1 and 2) was performed in a semi quantitative way on a

Shimadzu QP-2010 Gas Chromatograph interfaced to a mass spectrometer (GC-MS). The following conditions were used: DB-5MS column Agilent Technologies (30 m × 0.25 mm × 0.25 μm); helium (99.999 %) carrier gas at a constant flow of 1.1 ml/min; 1.0 μL injection volume; injector split ratio of 1:10; injector temperature 250 °C; electron impact mode at 70 eV; ion-source temperature 280 °C and transfer line temperature 260 °C. The oven temperature was programmed from 60 °C to 240 °C at a temperature ramp of 3 °C/min. A mixture of linear hydrocarbons (C₈H₁₈ – C₄₀H₈₂) was injected under the same experimental conditions that the samples and the identification of the constituents was performed by comparing the mass spectra obtained with those of the equipment database (Wiley 7 lib and NIST 08 lib), using the Kovats Index. Data were processed with help of the Shimadzu GC-MS Solution software. All 1D and 2D NMR spectra were obtained in a Bruker spectrometer Ascend™-400 model with 400 MHz (¹H) and 100 MHz (¹³C) and shifts (δ) in ppm. The samples were solubilized in DMSO-*d*₆, MeOD or CDCl₃ using TMS as internal standard.

2.2. Plant material

For isolation of compounds from *Leonotis nepetifolia*, the plant material (leaves) was collected in March 2014 and September 2016. For extraction of essential oil, the leaves were collected in January of 2015. All specimens were collected in Petrolina, state of Pernambuco, Brazil. The specimens were compared with the voucher (#5266), deposited at Herbário Vale do São Francisco (HVASF) in Universidade Federal do Vale do São Francisco. All procedures for access to genetic patrimony and associated traditional knowledge were carried out and the project was registered in SisGen (Register #A7F117F and #A459D5B).

2.3. Preparation of extracts

The leaves were dried in an oven at 40 °C during 72 h. Then, the plant material was powdered and subjected to maceration with ethanol 95 %. The extractive solution was concentrated under vacuum on a rotatory evaporator at 50 °C, yielding 8.039 g of the crude ethanol extract (CEE-Ln-2014). The fractions were obtained by liquid chromatography under vacuum with hexane, chloroform, ethyl acetate and methanol in ascending order of polarity.²⁰ For the leaves collected in 2016, the plant material was also dried in an oven at 40 °C during 72 h and powdered in a mill. Then, it was extracted using ultrasonic apparatus with hexane, chloroform, ethyl acetate and methanol in ascending order of polarity.

2.4. Extraction of essential oil

The leaves (745.0 g) were mashed and extracted by hydrodistillation for 2 hours in a Clevenger modified apparatus with petroleum ether as facilitator solvent drug.¹⁴ The essential oil was refrigerated and separate from the water phase. The percent of extractions were calculated as relationship of mass of essential oil by mass of fresh leaves from the mass of the essential oil obtained.

2.5. Isolation of constituents

The chloroform fraction (CF-Ln-2014) (900 mg) was subjected to a column chromatography with silica gel as stationary phase and eluted with hexane, chloroform and methanol individually or as binary mixture, yielding 232 fractions. The fractions were grouped in 29 groups, and the fractions 1-38 and 39-40 were codified as compounds **1** (36.4 mg) and **2** (37.1 mg), respectively.

The ethyl acetate fraction (AcF-Ln-2014) (3.67 g) was chromatographed over a silica gel

column eluted with chloroform, ethyl acetate and methanol, individually or as binary mixture yielding 19 fractions. The solid observed in the fraction 17 was washed with chloroform and identified as compound **3** (5.0 mg). The rest of fractions were grouped and the fraction 14-15 was subjected to silica gel column chromatography eluted with hexane, ethyl acetate and methanol, individually or as binary mixture yielding 163 fractions. The solid of the fraction 15 was successively washed with chloroform and identified as compound **4** (27.8 mg).

The second ethyl acetate fraction (AcF-Ln-2016) (1.59 g) also was chromatographed over a silica gel column and eluted with chloroform, ethyl acetate and methanol, individually or as binary mixture yielding 41 fractions. The fractions 33-37 were grouped and chromatographed by exclusion molecular using Sephadex® as stationary phase, methanol as eluent and at end 49 fractions was obtained. The subfractions 24-26 and 27-29 were grouped and identified as compound **5** (36.8 mg) and **6** (31.2 mg) respectively.

Hentriacontane (1): White, crystalline solid; C₃₁H₆₄; MW 436.00 (g/mol); EI-MS *m/z*: 141, 113, 99, 85, 71 and 57 (base peak). ¹H NMR (400 MHz, CDCl₃) δ_H (ppm): 0.84-1.36; ¹³C NMR (100 MHz, CDCl₃) δ_C (ppm): 14.11, 22.70, 29.37, 29.71, 31.94.

Phytol palmitate (2): Colorless, crystalline solid; C₃₆H₇₀O₂; MW 534.00 (g/mol); EI-MS *m/z*: 278, 123, 95, 68, 57 (base peak). ¹H NMR (400 MHz, CDCl₃) δ_H (ppm): 2.3 (2H, t), 4.59 (1H, d, *J* = 6.98), 5.31 (1H, t); ¹³C NMR (100 MHz, CDCl₃) δ_C (ppm): 14.12 (C-16'), 16.38 (C-3a), 19.72 (C-7a), 19.76 (C-11a), 22.63 (C-16), 22.70 (C-5), 22.72 (C-15a), 22.84 (C-13), 24.48 (C-14), 25.05 (C-12), 27.99 (C-15), 29.1-29.71(C-3'-15'), 31.94 (C-10), 32.69 (C-7), 32.81 (C-11), 34.43 (C-2'), 37.32 (C-8), 37.39 (C-6), 37.44 (C-9), 39.87 (C-4), 61.20 (C-1), 118.22 (C-2), 142.59 (C-2), 173.94 (C-1').

Stigmasteryl glucoside (3): White, amorphous solid; C₃₅H₅₈O₆; MW 574.00 (g/mol); ¹H NMR (400 MHz, CDCl₃) δ_H (ppm): 3.04 (1H, m); 3.64 (2H, dd); 4.21 (H1, d); 5.04

(1H, dd; $J = 8.4$ Hz); 5.18 (1H, dd; $J = 8.4$ Hz); 5.31 (1H, m); ^{13}C NMR (100 MHz, CDCl_3) δ_{c} (ppm): 11.63 (C-18), 11.74 (C-29), 19.06 (C-19), 19.60 (C-27), 20.55 (C-11), 20.80 (C-21), 21.06 (C-26), 23.82 (C-15), 25.42 (C-28), 28.44 (C-16), 30.69 (C-2), 31.28 (C-25), 31.33 (C-7), 31.38 (C-8), 36.18 (C-10), 36.79 (C-1), 38.28 (C-4), 39.20 (C-12), 39.70 (C-20), 41.82 (C-13), 49.57 (C-9), 50.54 (C-24), 55.31 (C-17), 56.22 (C-14), 76.80 (C-3), 121.16 (C-6), 128.80 (C-23), 137.98 (C-22), 140.43 (C-5).

6,7-dimethoxy-5,3',4'-trihydroxyflavone

(4): Yellow, amorphous solid; $\text{C}_{17}\text{H}_{14}\text{O}_7$; LC-ESI-MS/MS [M+H] 331.05, MS^2 186.16; ^1H NMR (400 MHz, DMSO) δ_{H} (ppm): 3.71 (3H, s, OCH_3 -6), 3.91 (3H, s, OCH_3 -7); 6.71 (1H, s, H-3), 6.87 (1H, H-8), 6.88 (1H, d, $J = 8.00$, H-5'), 7.42 (1H, d, $J = 2.11$, H-2'), 7.44 (1H, dd, $J = 2.11$ and 8.00, H-6'), 12.92 (OH-12). ^{13}C NMR (100 MHz, DMSO) δ_{c} (ppm): 56.76 (OCH_3 -7), 60.40 (OCH_3 -6), 91.82 (C-8), 103.15 (C-3), 105.53 (C-10), 113.92 (C-2'), 116.39 (C-5'), 119.47 (C-6'), 121.93 (C-1'), 132.36 (C-6), 146.25 (C-3'), 150.30 (C-4'), 152.59 (C-5), 153.08 (C-9), 159.074 (C-7), 164.75 (C-2), 182.58 (C-4).

Apigenin 7-O-glucoside (5a): Green, amorphous solid; $\text{C}_{21}\text{H}_{20}\text{O}_{10}$; LC-ESI-MS/MS [M+H] 433.10, MS^2 : 271.02; ^1H NMR (400 MHz, DMSO) δ_{H} (ppm): 3.19 (1H, t, H-4''), 3.28 (1H, m, H-2''), 3.31 (1H, m, H-5''), 3.43 (1H, m, H-3''), 3.46 (1H, m, H-6''), 3.70 (1H, d, H-6''), 5.08 (1H, d, $J = 3.3$ Hz, H-1''), 6.44 (1H, H-6), 6.84 (1H, d, $J = 2.09$ Hz, H-6), 6.88 (1H, s, H-3), 6.96 (2H, d, $J = 8.9$, H-3' and H-5'), 7.97 (2H, d, $J = 8.9$ Hz, H-2' and H-6'), 13.02 (OH-12). ^{13}C NMR (100 MHz, DMSO) δ_{c} (ppm): 60.39 (C-6''), 69.34 (C-4''), 72.96 (C-2''), 76.29 (C-5''), 77.01 (C-3''), 94.64 (C-6), 99.36 (C-8), 99.69 (C-1''), 102.83 (C-3), 105.17 (C-10), 115.92 (C-3' and C-5'), 120.75 (C-1'), 128.48 (C-2' and C-6'), 156.80 (C-5), 160.98 (C-9), 161.40 (C-4'), 162.80 (C-7), 164.40 (C-1), 181.90 (C-4).

Luteolin-7-O-glucoside (5b): Green, amorphous solid; $\text{C}_{21}\text{H}_{20}\text{O}_{11}$; LC-ESI-MS/MS [M+H] 449.07, MS^2 : 287.03; ^1H NMR (400 MHz, DMSO) δ_{H} (ppm): 3.19 (1H, t, H-4''), 3.28 (1H, m, H-2''), 3.31 (1H, m, H-5''), 3.43 (1H, m, H-3''), 3.46 (1H, m, H-6''), 3.70 (1H, d, H-6''), 5.08 (1H, d, $J = 3.3$ Hz, H-1''), 6.44 (1H, m, H-6), 6.76 (1H, s, H-3), 6.79 (1H, d, $J = 2.09$ Hz, H-6),

6.92 (1H, d, $J = 9.00$ Hz, H-3'), 7.43 (2H, d, $J = 9.00$ Hz, H-2'), 7.45 (1H, s, H-6'), 13.02 (OH-12). ^{13}C NMR (100 MHz, DMSO) δ_{c} (ppm): 60.39 (C-6''), 69.34 (C-4''), 72.96 (C-2''), 76.29 (C-5''), 77.01 (C-3''), 94.55 (C-6), 99.36 (C-8), 99.69 (C-1''), 102.83 (C-3), 105.17 (C-10), 113.44 (C-6'), 115.92 (C-3'), 119.06 (C-2'), 120.75 (C-1'), 145.71 (C-5'), 150.41 (C-4'), 156.80 (C-5), 160.98 (C-9), 162.80 (C-7), 164.30 (C-1), 181.90 (C-4).

Luteolin-7-O-glucoside (6): Green, amorphous solid; $\text{C}_{21}\text{H}_{20}\text{O}_{11}$; LC-ESI-MS/MS [M+H] 449.07, MS^2 : 287.03; ^1H NMR (400 MHz, MeOD) δ_{H} (ppm): 3.41 (1H, m, H-3''), 3.49 (1H, m, H-2''), 3.53 (1H, m, H-4''), 3.55 (1H, m, H-5''), 3.75 (1H, d, $J = 3.3$ Hz, H-6''), 3.93 (1H, d, $J = 3.3$ Hz, H-6''), 5.07 (1H, d, H-1''), 6.49 (1H, d, $J = 1.98$, H-8), 6.61 (1H, s, H-3), 6.80 (1H, d, $J = 1.98$ Hz, H-6), 6.91 (1H, d, $J = 8.78$ Hz, H-3'), 7.41 (1H, d, $J = 8.78$ Hz, H-2'), 7.43 (1H, s, H-6'). ^{13}C NMR (100 MHz, DMSO) δ_{c} (ppm): 62.54 (C-6''), 71.36 (C-3''), 74.84 (C-2''), 77.9 (C-4''), 78.5 (C-5''), 95.96 (C-6), 100.87 (C-8), 101.50 (C-1''), 103.8 (C-3), 107.08 (C-10), 114.28 (C-2'), 116.65 (C-3'), 120.42 (C-6'), 123.49 (C-1'), 147.15 (C-5'), 151.25 (C-4'), 159.27 (C-5), 162.93 (C-9), 164.84 (C-7), 166.97 (C-2), 184.14 (C-4).

2.6. Cytotoxic activity

Human tumor cell lines, HCT-116 (human colon) and SF-295 (glioblastoma) were obtained from the National Cancer Institute (Bethesda, MD, USA). All cells were maintained in RPMI 1640 medium supplemented with 10 % fetal bovine serum, 100 U/ml of penicillin, and 100 $\mu\text{g}/\text{ml}$ of streptomycin at 37 °C with 5 % CO_2 . The essential oil and compounds were tested for cytotoxic activity against the tumor cell lines.

Cells were plated at concentrations of 0.7×10^5 and 0.1×10^6 cells/mL for HCT-116 and SF295 strains, respectively. The plates were incubated with the sample for 72 hours in an incubator at 5 % CO_2 and 37 °C. At the end, they were centrifuged, and the supernatant removed. Then, it was added 150 μL solution

of 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide (MTT) and the plates were incubated for 3 h. After incubation, the plates were centrifuged again to remove the MTT solution. The absorbance was read after dissolution of the formazan precipitate with

150 μ L pure DMSO in a plate spectrophotometer (595 nm). The cytotoxic effect was quantified as the percentage of the control absorbance of the reduced dye at 595 nm.²¹ All absorbance values were converted into cell growth inhibition (GI) values, using:

$$\text{Eq 1.} \quad GI(\%) = 100 - [(T/C)100]$$

Where, **C** was the absorbance for negative control and **T** was the absorbance in the presence of the tested sample.

Brazil and Nigeria, respectively, are responsible by differences on oil constitution of Brazilian and Nigerian specimens.^{7,13}

3. Results and Discussion

3.1. Characterization of essential oil

Light green oil was distilled from the leaves of *L. nepetifolia*. A total of 33 compounds were found from the GC-MS analysis and 93.94 % (31 compounds) were identified by comparison of the data with device libraries (Table 1). In this sample, more than 60.0 % of the identified compounds presented up to 9 carbons in its structures which explains the high volatility of oil. The compounds germacrene D, α -humulene, α -pinene, 3-octanone, 1,8-cineole, (*E*)-ocimene, (*Z*)-ocimene, linalool, caryophyllene oxide and the major compound 1-octen-3-ol present in the essential oil from the *L. nepetifolia*, were also reported in the literature for this species and in others Lamiaceae species.^{7,13, 22, 23}

The essential oil from the leaves of Nigerian *L. nepetifolia* specimen showed the compounds β -caryophyllene and α -humulene as majoritary constituents and the 1-octen-3-ol as traces. The differences on geography and climate of countries, Caatinga and Savana for

3.2. Isolated constituents

The structures of the isolated compounds (Figure 1) were confirmed by appropriate spectroscopic methods such as ¹H NMR, ¹³C NMR, GC-MS and LC-ESI-MS. The ¹H NMR spectra of compound **2** showed a triplet at 5.31 ppm for the olefinic hydrogen and a doublet at 4.59 relative to oxymetilenic hydrogens typical of phytol skeleton. The ¹³C NMR spectra beyond the typical signal of olefinic carbons 118.2 ppm and 142.60 ppm typical of phytol skeleton was observed one signal at 173.94 ppm, suggesting the presence of one carbonyl ester in the structure.²⁴ Alkyl chain signals also were observed in the ¹³C NMR spectra and the HMBC spectra showed correlations between the signals at 4.68 ppm and 2.30 ppm with the carbonyl carbon (173.94 ppm) confirming the presence of a phytol ester in the structure. The size of the aliphatic chain was determined by GC-MS. The fragment *m/z* 278 was attributed to the phytol structure and the difference of total mass *m/z* 534 with NMR data allowed getting at the structure of 3,7,11,15-tetramethylhexadec-2-enyl palmitate, isolated for the first time in the *L. nepetifolia* species.

Table 1. Chemical composition of the essential oil from the leaves of *Leonotis nepetifolia*

Peak	RT (min)	Compound	(%)
1	4.099	o-Xylene	2.35
2	4.139	4-Methyl-hexanal	1.31
3	4.551	m-Xylene	3.05
4	4.610	Heptaldehyde	0.12
5	4.673	Nonane	0.57
6	4.807	Hexa-2,4-dienal	0.65
7	5.151	Ethyl disulfide	0.28
8	5.543	α -Pinene	0.46
9	5.683	4-Methyl-hexanol	0.51
10	6.507	2-Methyl-3,4-dithiahexane	0.92
11	6.670	1-Octen-3-ol	42.58
12	6.771	3-Octanone	3.75
13	7.006	Fenchone	0.75
14	7.171	Octan-3-ol	1.58
15	7.272	(2-Pentenyl)furan	0.51
16	7.373	(Z)-3-Hexenyl acetate	1.28
17	7.929	Isopropyl disulfide	0.61
18	8.391	1,8-Cineole	0.73
19	8.515	(E)-Ocimene	15.85
20	8.896	(Z)-Ocimene	7.01
21	9.289	3,7-Dimethyl-undecane	0.59
22	10.835	Linalool	2.05
23	12.022	2,6-Dimethyl-2,4,6-octatriene	0.59
24	18.950	Dihydroedulan I	0.41
25	22.515	β -Damascenone	0.92
26	23.104	β -Elemene	1.06
27	24.297	β -Caryophyllene	3.22
28	25.722	α -Humulene	1.82
29	26.664	β -Ionone	0.83
30	26.783	Germacrene D	2.41
31	30.672	Caryophyllene oxide	0.72
32	40.489	NI	0.39
33	59.494	NI	0.10

RT: retention time (minutes); (%) percentage of compound on the sample; NI: Not identified

The ^1H NMR spectra of compound **3** showed signals among 0.48-2.90 ppm for methyl and methylene hydrogens, two double doublets at 4.98 ppm and 5.1 ppm corresponding to olefinic hydrogens, a distorted triplet at 5.30 ppm typical of steroid

skeletons with double bond in C-5, and the signal at 4.20 ppm characteristic of the one anomeric proton with $J = 8.0$ Hz, suggesting the presence of one sugar unit with β -configuration.²⁵

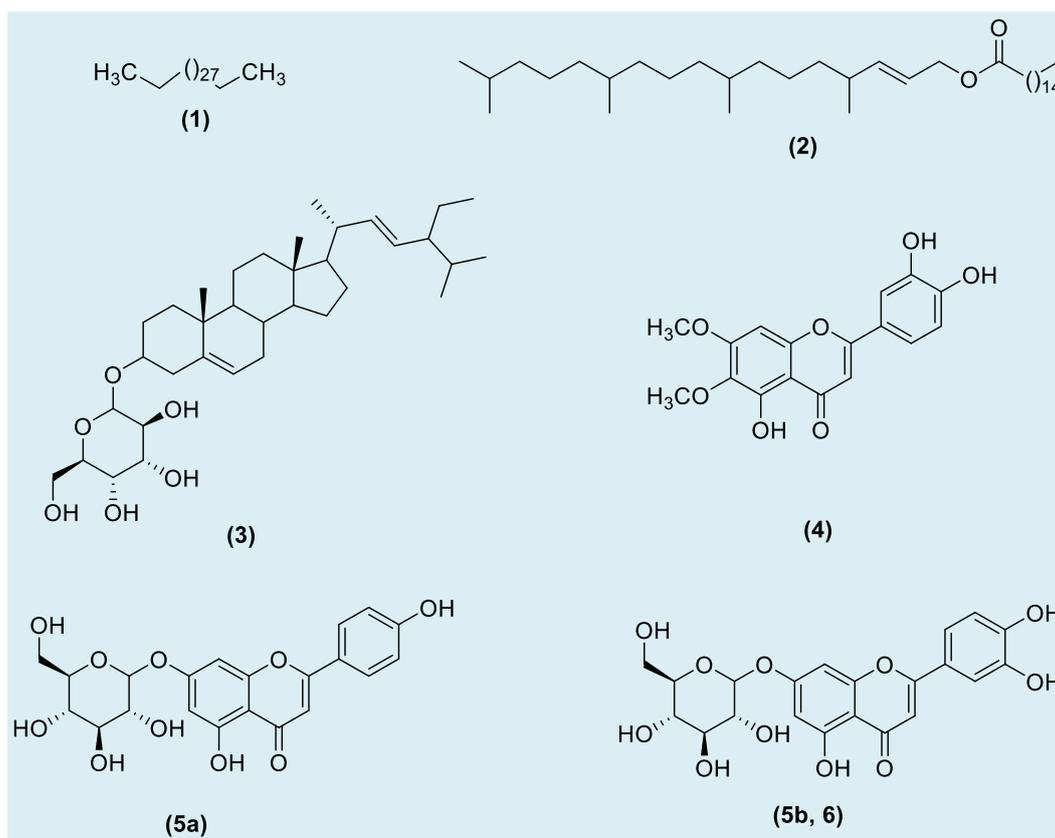


Figure 1. Compounds isolated from the leaves of *Leonotis nepetifolia*

The ^{13}C NMR spectra showed signals characteristic of one sugar unit as the signal at 101.75 ppm, typical of anomeric carbons beyond the six signals in the glycoside region (61.0-77.00 ppm). The olefinic carbons signals at 121.16 and 140.43 ppm, at 128.79 and 137.98 ppm related the double bond between C5-C6 and C22-C23, respectively, confirmed the stigmasteryl skeleton.²⁵ The correlations between hydrogen at 4.21 ppm of the sugar unit with C-3 of the aglycone at 76.88 ppm showed the linkage of sugar moiety in this carbon.²⁵

The MS and NMR data of compound **4** are the same as reported in the literature for 6,7-dimethoxy-5,3',4'-trihydroxyflavone (cirsiolol), a flavonoid already reported for *L. nepetifolia*.^{10,20} The ^1H NMR spectra of compound **5** showed a mixture of two flavonoids with typical multiplicity for rings B of flavonoids 1,4-substituted and 1,3,4-substituted, respectively.^{26,27} The comparison of all spectroscopic and spectrometric data

with literature,^{26,27} allowed affirming the presence of flavonoids apigenin-7-*O*-glucoside and luteolin-7-*O*-glucoside in mixture and still that the compound **6** it was the flavonoid luteolin-7-*O*-glucoside.

3.3. Cytotoxic activity

Cytotoxicity analysis by the MTT method is commonly used. Large reference centers have used this method, such as the National Cancer Institute (NCI), which tests over 10.000 substances each year.²⁸ It is a fast, sensitive and inexpensive method. It was first described by Mosman (1983) and allows analyzing the viability and the metabolic state of the cell. It is a colorimetric analysis based on conversion of the salt 3-(4,5-dimethyl-2-thiazole)-2,5-diphenyl-2-H-tetrazolium bromide (MTT) into formazan blue, by mitochondrial enzymes present only in metabolically active cells. The cytotoxic study by MTT method allows easily

set the cytotoxicity but not the mechanism of action.²⁹

A previous study showed the high potential cytotoxic of *L. nepetifolia* extracts and the compound 6,7-dimethoxy-5, 3',4'-trihydroxyflavone (cirsiol) isolated in this

study.²⁰ The essential oil, hentriacontane, phytol palmitate, stigmasteryl glucoside, the mixture of flavonoids apigenin-7-*O*-glucoside and luteolin-7-*O*-glucoside and luteolin-7-*O*-glucoside alone, showed weak activity for the tested strains (Table 2).⁶

Table 2. Cell proliferation inhibition (CI%) of essential oil and chemical compounds of *Leonotis nepetifolia*

	HCT-116	SF-295
EO	16.78 ± 0.43	13.06 ± 11.06
C1	26.41 ± 1.37	00.00 ± 00.00
C2	32.78 ± 4.89	12.32 ± 14.38
C3	47.02 ± 4.96	23.15 ± 10.19
C5(5a/5b)	13.37 ± 0.92	00.000 ± 00.00
C6	11.25 ± 2.54	2.74 ± 3.83

Mean of percentage of cell growth inhibition (CI%) of samples at 50 µg.mL⁻¹. Values expressed as mean ± standard error of mean with n = 3. (EO) essential oil, (C1) Hentriacontane, (C2) Phytol palmitate, (C3) Stigmasteryl glucoside, (C5), apigenin-7-*O*-glucoside and luteolin-7-*O*-glucoside, (C6) luteolin-7-*O*-glucoside. (HTC 116) human colon and (SF295) glioblastoma

A study developed with 1-oct-3-enol, major compound of essential oil from the leaves of *L. nepetifolia*, revealed a strong cytotoxic activity against human embryonic stem cells.³⁰ Although it appears contradictory, the discrepancies between the results of the present study and the realized by Inamdar et al (2012), can be explained by kind of treatment used in the tests and by adopted concentration. In the above study, the authors adopted the airborne exposure technique and a 0-1000 ppm as scale concentration for 1-oct-3-enol while in our study a less (50 µg/mL) and single concentration was used.³⁰

Previous reports have showed that stigmasteryl aglycone exhibited a weak antiproliferative activity against hepatic liver (WRL), breast carcinoma (MCF-7), colorectal carcinoma (COLO), T-lymphoblastic leukemia cell line (CEM), cervical carcinoma cell line HeLa and human fibroblasts (BJ) cell lines, at the concentrations of 50 and 100 µM.³¹⁻³² These and ours results show that the sugar

unit was not able to potentiate the cytotoxic activity of the stigmasteryl molecule.

The presence of flavonoids apigenin-7-*O*-glucoside and luteolin-7-*O*-glucoside in species *L. nepetifolia* it's very interesting, although our results have shown a weak cytotoxic activity for mixture of flavonoids and for luteolin-7-*O*-glucoside, because, these flavonoids are metabolized in its respective aglycones apigenin and luteolin,³³ whose *in vitro* cytotoxic activity against tumor line cells³⁴⁻³⁵ and *in vivo* antitumor activity³⁵ have been proven.

4. Conclusion

Our study describes for the first time the isolation of hentriacontane, phytol palmitate, stigmasteryl glucoside, apigenin-7-*O*-glucoside and luteolin-7-*O*-glucoside in *Leonotis nepetifolia*, and the cytotoxic activity for these compounds and essential oil in

human tumor cell lines. In this context, this paper contributes for the phytochemical study of this vegetal species as well as for the Lamiaceae family.

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