Artigo

Triterpenes esterified with fatty acid isolated from *Pouteria*macahensis T.D. Penn (Sapotaceae) leaves

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Triterpenos esterificados com ácidos graxos isolados das folhas de *Pouteria macahensis* T.D. Penn (Sapotaceae)

Resumo: *Pouteria macahensis* é uma espécie endêmica da Mata Atlântica, de ocorrência em áreas restritas. Nesse trabalho é apresentado o estudo fitoquímico do extrato etanólico das folhas de *P. machaensis*, coletadas em Itacaré, Bahia, Brasil. Folhas secas e trituradas foram submetidas à extração exaustiva com hexano e em seguida etanol. O estudo do extrato etanólico permitiu a identificação de seis triterpenos pentacíclicos esterificados com ácidos graxos: mistura de behenato e erucato de 3β-erithrodiol, palmitoleato de 3β-erithrodiol e mistura de behenato, erucato e palmitoleato de ácido 3β-betulínico. As identificações estruturais foram baseadas nas técnicas espectroscópicas de IV, RMN de ¹H e ¹³C (1D e 2D) e dos ésteres de ácidos graxos, após reação de transesterificação, através da análise por cromatografia a gás. A presença desses compostos pode estar associada à localização das espécies na Mata Atlântica, em terreno arenoso e com grande incidência de sol.

Palavras-chave: Mata Atlântica, Lacuna macahensis, transesterificação.

Abstract

Pouteria macahensis is an endemic species of the Atlantic Rainforest, which occurs in restricted regions. This work presents the phytochemical study of the leaf ethanolic extract from *P. macahensis*, collected in Itacaré, Bahia, Brazil. Dry and ground leaves were submitted to exhaustive extraction in hexane, followed by ethanol. The ethanolic extract yielded six fatty acid esters of pentacyclic triterpenes: a mixture of 3β -erythrodiol behenate and erucate; 3β -erythrodiol palmitoleate; and the mixture of 3β -betulinic acid behenate, erucate, and palmitoleate. The structural identifications were based on the spectroscopic techniques of IR, 1 H, and 13 C NMR (1D and 2D), as well as the esters of fatty acids, after the reaction of transesterification and gas chromatography analysis. The presence of these compounds can be associated with the localization of the species in the Atlantic Rainforest, in a sandy terrain and in a region with constant sunshine.

Keywords: Atlantic Rainforest, Lacuna macahensis, transesterification.

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Triterpenes esterified with fatty acid isolated from *Pouteria* macahensis T.D. Penn (Sapotaceae) leaves

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1. Introduction

Pentacyclic triterpenes, such as friedelanes, oleananes, ursanes, and lupanes are commonly isolated in plant species; however, triterpenes esterified with long chain fatty acids are rarer. Triterpenes are considered to be chemotaxonomic markers from the Sapotaceae family, which consists of 53 genera and approximately 1,100 species, distributed in tropical and subtropical regions.¹

Ethnopharmacological studies of the Sapotaceae family indicate that the *Pouteria*

genus is among the most widely studied.² This genus, which is most commonly found in South America, is comprised of 122 species, which 50 are endemic to the Atlantic Rainforest. In Brazil, there are 11 genera and 231 species distributed throughout the states of Amazônia, Bahia, Espírito Santo, Rio de Janeiro, and São Paulo.³

In the genera of *Gambeya, Pradosia*, and *Manilkara* (*Sapotaceae* family), triterpenes esterified with the acetyl group and with long chain fatty acids were isolated; however, some researchers no identify the long chain. Erythrodiol, β and α -amyrin esters were isolated from *G. africana* fruits, *M. bidentate* and *M. subsericea* (Mart).



of *Pradosia huberi* (Ducke) were isolated long chain fatty acids derived from erythrodiol and betulinic acid.⁷

The *Pouteria* genus, in the bark of *P. campechiana*, yielded esters derived from erythrodiol, oleanolic acid, and betulinic acid.⁸ From the *P. torta* leaves, palmitate and erythrodiol were isolated.⁹

Pouteria macahensis T.D. Penn (a synonym for Lacuma macahensis Glaz), popular name "Maçaranduba-mirim", is an endemic tree to the Atlantic Rainforest, distributed along the Brazilian coastline in the states of Alagoas, Bahia, Espírito Santo, and Rio de Janeiro.³ Due Pouteria to be a genus of wood plants, the occurrence of this species is more and more scarce, especially in urban centers and its surroundings. The present study offers a contribution to the chemical study of the P. macahaensis species, through the isolation of substances from its leaf ethanolic extract. To the best of our knowledge, this is the first chemical study conducted with the leaves of this species.

2. Materials and Methods

2.1 General Procedures

For Thin Layer Chromatography (TLC) and Preparative Thin Layer Chromatography (Prepative TLC) were used silica gel Vetec® 60-G, and to Column Chromatography (CC) were used silica gel Merck® 60 (0.063 – 0.200 mm). Spectra of de ¹H and ¹³C NMR (1D and 2D) were acquired in the Bruker Avance® CRX400 device, using TMS as a reference. The IR spectra were recorded in the Perkin Elmer® 400 device, and melting points were recorded in the Fisatom® 430D Capillary Fusion Point without correction. To characterize the fatty acid esters chain bonded to triterpenes, the transesterification methodology was adopted, according to that proposed by Mendes. 10 The analysis of the fatty acid esters was performed the Varian Saturm 3800 gas chromatography (GC), equipped with FID, Carbowax fused-silica capillary column (30m x 0.25mm x 0.25 μ m), He 5.0 with a flow of 1.0 mL min ⁻¹, and temperatures: injector 220 °C, detector 240 °C. The initial temperature of the column was 60 °C, raised by 5 °C min⁻¹ until reaching 200 °C, and was then maintained at this temperature for five minutes. The FAME mixture, C₈-C₂₂, (Supelco – USA; catalog 47801- Lot: LC00126) was used to identify the chemical composition of the esters through the comparison of the retention times.

2.2 Plant material

P. macahensis was collected in the town of Itacaré, Bahia, Brazil, in September 2015 (coordinates 40°15'53.0"S 38°59'59.6"W) and was identified by Luiz Alberto Mattos Silva and José Lima da Paixão. The specimen voucher was deposited in the Herbarium of the State University of Santa Cruz, logged under protocol number 21,469, and the leaves were dried in a forced ventilation oven.

2.3 Extraction and isolation

The plant material (2100 g) was submitted to exhaustive maceration in hexane and ethanol at room temperature. The ethanolic extract (80 g) was chromatographed in CC, using as eluents Hex, CHCl₃, AcOEt, and EtOH (pure and in mixtures), yielding 34 fractions. After TLC analyses (Hex, CHCl₃ 4:6), the samples were joined in 15 groups (G1-G15). Group G4 (1.9767 g) was treated with acetone, resulting in a solid; the remainder of the material (1.4567 g) was chromatographed in CC using the eluents Hex and CHCl₃ (pure and in mixtures), supplying 21 fractions, grouped by TLC (DCM, CHCl₃ 2:8) in 4 groups (A-D). Group B (0.1820 g) was treated with acetone and recrystallized with methanol, supplying a mixture of compounds 1a-b (0.0215g). This mixture of compounds 1a-b was transesterified with MeOH/MeONa, leading to substance 2 (7 mg). The filtrated group B was submitted to preparative TLC, using CH₂Cl₂/CHCl₃ (8:2), allowing for the



isolation of compound **3** (0.0267 g). Group G6 (1.2342 g) was treated with acetone and recrystallized with MeOH, resulting in a mixture of compounds **4a-c** (0.0291g). The mixture of compounds **4a-c**, after transesterification, resulted in compound **5** (8 mg).

2.4. Spectroscopic data

*3*β-erythrodiol behenate and erucicate (**1a-b**): Amorphous white solid (MeOH), M.p.: 118-121 °C; IR (KBr, cm⁻¹): 3474, 2952, 2863, 1735, 1465, 1384. ¹H NMR (400 MHz, CDCl₃) δ: 0.86 (6H, s, CH₃-29 and CH₃-30); 0.87 (3H, s, CH₃-24); 0.89 (3H, s, CH₃-23); 0.94 (3H, s, CH₃-25); 0.95 (3H, s, CH₃-26); 1,16 (3H, s, CH₃-27); 1.25 (nH, s, H-C(n-fatty acid); 1.55-1.98 (nH, m, H-C(n-alkyl); 2.28 (2H, t, t) = 7.4Hz, H-2'); 3.21 (1H, t), t0.9Hz, H-28b); 4.5 (1H, t0, t1, t3.2 Hz, H-12); 5.29 (1H, t5, t5.17 (1H, t7, t7, t8.3 Hz, H-12); 5.29 (1H, t8, t9, H-9'). ¹³C NMR (100 MHz, CDCl₃): Table 1.

3β-erythrodiol (**2**): White powder (EtOH), M.p.: 228-230 °C. IR (KBr, cm⁻¹): 3492; 2925; 2855; 1465; 1352. ¹H NMR see Wang *et* al. ¹¹ ¹³C NMR: Table 1.

3β-erythrodiol palmitoleate (3): Amorphous white solid (MeOH), M.p: 122-124 °C. IR (KBr, cm⁻¹): 3478, 2930, 2875, 1463, 1383, 1168. ¹H NMR: Similar data to

compounds **1a-b**. ¹³C NMR (100 MHz, CDCl₃): Table 1.

 3β -betulinic acid behenate, erucicate and palmitoleate (**4a-c**): White waxy. IR (KBr, cm⁻¹): 3437, 2940, 2848, 1725, 1456, 1374, 1065; ¹H NMR (400 MHz, CDCl₃) δ: 0.89(3H, s, CH₃); 0.83 (3H, s, CH₃-25); 0.84 (3H, s, CH₃-24); 0.85 (3H, s, CH₃-23); 0.94 (3H, s, CH₃-26); 0.98 (3H, s, CH₃-27); 1.69 (3H, s, CH₃-30); 1.25 (nH, s, H-C(n-fatty acid); 1.55-1.98 (nH, m, H-C(n-alkyl); 2.28 (2H, J = 7.5 Hz); 3.09 (1H, m, H-19); 4.46 (1H, dd, J = 5.8 Hz and 10.2, Hz, H-3); 4.61 (1H, br, H-29a); 4.73 (1H, br, H-29b). ¹³C NMR (100 MHz, CDCl₃): Table 1.

Betulinic acid (5): White solid; M.p:295-298 $^{\circ}$ C (dec.). IR (KBr, cm⁻¹): 3060, 1630, 880, 1 H NMR see Haque et al. 12 13 C NMR: Table 1

3. Results and Discussion

The phytochemical study of the leaf ethanolic extract led to the identification of six esterified triterpenes **1a-b**, **3**, and **4a-c**, while compounds **2** and **5** were obtained after the hydrolysis of the esterified triterpenes (Figure 1).

1a: R:
$$-CO + CH_2 + CH_2 + CH_3$$

1b: R: $-CO + CH_2 + CH_3 + CH_2 + CH_3 + CH_2 + CH_3 + C$

4a: R:
$$-CO + CH_2 + CH + CH_2 + CH_3$$

4b: R: $-CO + CH_2 + CH + CH_2 + CH_3$
4c: R: $-CO + CH_2 + CH_3 + CH_3 + CH_3$
5: R: H

Figure 1. Chemical structures of compounds 1-5



The structural identifications were and 13 C NMR (Table 1), DEPT-135, HSQC, and conducted through the analyses of the IR, 1 H, HMBC spectra.

Table 1. ¹³C NMR (100 MHz) chemical shifts of compounds 1-5

Carbon	δ _c (1a-b) ^a	δ _c (2) ^b	δ _c (3) ^a	δ _c (4a-c) ^a	δ _c (5*) ^b
1	38.28	38.78	38.24	38.43	39.31
2	23.81	23.5	23.53	28.30	28.33
3	80.55	79.02	80.54	80.61	78.10
4	37.76	37.75	37.75	38.41	39.54
5	55.27	52.21	55.22	55.46	56,00
6	18.25	18.25	18.23	18.18	18.83
7	32.54	32.60	32.49	34.28	34.91
8	39.82	39.81	39.79	40.74	41.11
9	47.52	47.60	47.48	50.44	51.08
10	36.84	36.95	36.81	37.15	37.52
11	23.54	23.54	23.59	20.88	21.23
12	122.31	122.38	122.29	25.48	26.17
13	144.22	144.21	144.22	38.35	38.60
14	41.74	41.75	41.71	42.45	42.92
15	25.55	25.56	25.52	30.58	30.31
16	22.03	22.04	21.96	32.17	32.94
17	36.95	36.95	36.94	56.34	56.62
18	42.36	42.36	42.34	49.31	49.81
19	46.47	46.49	46.4	46.93	47.83
20	30.94	30.94	30.97	150.36	151.33
21	34.10	34.10	34.07	31.37	31.24
22	31.04	31.08	31.04	37.04	37.60
23	28.91	28.02	28.04	27.97	28.72
24	15.56	15.56	15.58	16.05	16.32
25	16.75	16.70	16.73	16.16	16.42
26	16.80	16.75	16.78	16.53	16.40
27	25.90	25.90	25.92	14.67	14.92
28	69.71	69.70	69.75	180.75	178.81
29	33.17	33.18	33.20	109.70	109.92
30	23.58	23.58	23.46	19.35	19.54
1'	173.67		173.76	173.71	
2'	34.86/34.82		34.07	34.81/34.86	
3'	25.18		25.2	25.18	
4'-9'	29.69		29.69	29.69	
10'	29.69		29.69	29.69	
11-13′	29.69		29.69	29.69	
14'	29.69		31.97	29.69	
15′	29.69		21.96	29.69	
16′	29.69		14.10	29.69	
17'-19'	29.69			29.69	
20′	31.92			31.92	
21'	22.57/22.67			22.57/22.67	
22′	14.03/14.09			13.88/14.02/14.08	

^a(CDCl₃), ^b(CDCl₃ and Py-d5), *(Py-d5) literature.



The ¹H NMR spectrum of the mixture of compounds 1a-b presented six singlets between δ_{H} 0.86 and 9.95, characteristic of the seven methyl groups; an intense singlet at δ_{H} 1.25, characteristic of alkyl chain methylene hydrogen, and a triplet at δ_H 2.28 (2H, J = 7.8 referent to the H-2' hydrogen, neighboring C=O of the triterpene ester. The signal at δ_H 4.51 (1H, m, H-3) was attributed to the oxymethinic hydrogen at C-3. The presence of two defined doubles at δ_{H} 3.21 (1H, d, J = 10.9 Hz) and $\delta_H 3.54$ (1H, d, J = 10.9Hz), characteristic of coupled oxymethylene hydrogens and a triplet at δ_H 5.17 (1H, t, J = 3.2 Hz), suggests the presence of the erythrodiol nucleus.¹¹ correlations The H-C confirmed in the HSQC experiment. The correlations observed in the **HMBC** experiment (Figure 2) allow the assignment of the double bond at C-12 and C-13, through the correction of the signal at δ_H 5.17 with the signals at δ_{C} 47.52 (C-9), 23.54 (C-11), and 42.36 (C-18). The presence of the oxymethylene group at C-28 was confirmed through the correlation between δ_H 3.54 and δ_{H} 3.21 with δ_{C} 22.03 (C-16) and 31.04 (C-22). The correlation of the signal at δ_H 4.51 with δ_C 38.28 (C-1), 23.81 (C-2), 28.91 (C-23), 15.56 (C-24), and 173.67 (C-1') confirms the triterpene esterification at the C-3 position.

$$H_3C(H_2C)H_2C$$

1a-b

 $H_3C(H_2C)H_2C$
 H_3C

Figure 2. Main correlations observed for HMBC_s (H \rightarrow C) from compounds 1a-b and 4a-c

To identify the long chain fatty acid, the mixture of the compounds ${\bf 1a-b}$ was transesterified with MeOH/MeONa. The GC analysis of the CHCl3 phase of this reaction indicated the presence of methyl behenate and methyl erucate. Thus, the compounds ${\bf 1a-b}$ are a mixture of ${\bf 3\beta}$ -erythrodiol behenate and erucate. From the aqueous phase, a white solid was obtained (compound ${\bf 2}$); its structural elucidation was based on IR, 1 H NMR, 13 C NMR (proton decoupled and DEPT-135) analysis and by comparison with erythrodiol spectral data. 11

The IR, ¹H, and ¹³C NMR data (Table 1), DEPT-135, HSQC, and HMBC from compound **3** suggest the presence of esterified erythrodiol, which is similar to the mixture of compounds **1a-b.** In a similar manner, the

fatty acid chain was characterized through the reaction from the transesterification of this compound. The GC analysis indicated the presence of methyl palmitoleate. Thus, compound $\bf 3$ is $\bf 3\beta$ -erythrodiol palmitoleate. The mixture of compounds $\bf 1a$ - $\bf b$ and compound $\bf 3$ are being reported for the first time in the *Pouteria* genus. Esterified erythrodiol with long chain fatty acid (without the identification of the long chain) was isolated from *P. campechiana*, and from *P. torta* leaves was isolated erythrodiol palmitate.

The spectrum of ^1H NMR of the mixture of compounds ~4a-c~ presented six signals between δ_{H} 0.83 and 1.69, characteristics of the six methyl groups, an intense singlet at δ_{H}



1.25, characteristic of an alkyl chain methylene hydrogen and a triplet at δ_H 2.28 (2H, J = 7.5 Hz), equivalent to the H-2' hydrogen, neighboring the ester C=O. The signal at δ_H 4.46 (1H, dd, J = 5.8 and J = 10.2Hz) was attributed to H-3 of the oxymethinic hydrogen, suggesting a compound containing C-3 esterified. The presence of two broad singlets at δ_H 4.46 (1H) and δ_H 4.73 (1H) was assigned to terminal olefin hydrogens, suggesting the presence of a betulinic acid nucleus.¹² The main correlations observed in the HMBC experiment (Figure 2) confirmed the presence of the terminal double bond at C-29 and C-20, through the correlations of signals at δ_{H} 4.73 and δ_{H} 4.61 with δ_{C} 19.35, (C-30), 150.36 (C-20), and 46.93 (C-19). The correlation of the signal at δ_H 4.46 with signals at $\delta_{\rm C}$ 38.43 (C-1), 28.30 (C-2), 27.97 (C-23), 16.05 (C-24), and 173.71 (C-1') confirmed the esterification in the C-3 carbon. The correlation of the signal at δ_H 3.09 (1H, m, H-19), with $\delta_{\rm C}$ 19.35 (C-30), 31.37 (C-21), 49.31 (C-18), and 109.70 (C-29), confirmed the position of the olefin at C-20 and C-29. These chemical shifts are compatible with that reported for the esterified betulinic acid with the long chain fatty acid.8

The mixture of compounds **4a-c** was transesterified with MeOH/MeONa. The GC analysis indicated the presence of methyl behenate, methyl eurucate, and methyl palmitoleate.

Thus, the compounds **4a-c** are a mixture of 3β-betulinic acid behenate, erucate, and palmitoleate, which is being reported for the first time in this study. Ragasa et al. isolated a mixture of esterified betulinic acid with long chain acid from *P. campechiana*; however, the alkyl chain was not identified.⁸

From the aqueous phase of the transesterifictaion reaction, a solid was obtained, compound **5.** Its structural elucidation was based on the analyses performed in IR, ¹H, and ¹³C NMR spectra, as compared to assignments for the betulinic acid. ¹²

4. Conclusions

The extract from *P. macahensis* leaves is rich in triterpenes esterified with long chain fatty acids. This occurrence may well be associated with the localization of individual species in the Atlantic Rainforest, in a sandy terrain and in a region with constant sunshine.

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